

# Efficient Syntheses of Korupensamines A, B and Michellamine B by Asymmetric Suzuki-Miyaura Coupling Reactions

Guangqing Xu,<sup>‡</sup> Wenzhen Fu,<sup>‡</sup> Guodu Liu,<sup>‡</sup> Chris H. Senanayake,<sup>†</sup> and Wenjun Tang<sup>\*‡</sup>

<sup>‡</sup>State Key Laboratory of Bio-Organic and Natural Products Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, Shanghai 200032, China

<sup>†</sup>Chemical Development, Boehringer Ingelheim Pharmaceuticals Inc, Ridgefield, Connecticut CT06877, United States

**S** Supporting Information

**ABSTRACT:** Efficient asymmetric Suzuki-Miyaura coupling reactions are employed for the first time in total syntheses of chiral biaryl natural products korupensamine A and B in combination with an effective diastereoselective hydrogenation, allowing ultimately a concise and stereoselective synthesis of michellamine B. Chiral monophosphorus ligands L1–3 are effective for the syntheses of a series of functionalized chiral biaryls by asymmetric Suzuki-Miyaura coupling reactions in excellent yields and enantioselectivities (up to 99% ee). The presence of a polar- $\pi$  interaction between the highly polarized BOP group and the extended  $\pi$  system of arylboronic acid coupling partner is believed to be important for the high enantioselectivity.

Michellamine B (**1**),<sup>1</sup> originally isolated from tropical liana *Ancistrocladus korupensis* in Cameroon, has gained significant interest as a strong anti-HIV-1 and anti-HIV-2 agent (Figure 1). Structurally, it derives from heterodimerization of

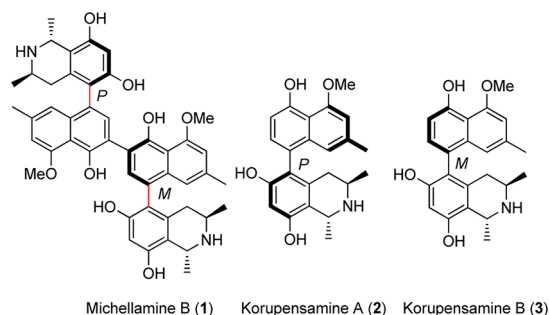


Figure 1. Axially chiral biaryl natural products.

antimalarial atropdiastereomeric natural products korupensamine A (**2**) and korupensamine B (**3**).<sup>2</sup> Due to their interesting biological activities and unique molecular architectures containing both axial and central chirality, there has been considerable study<sup>3</sup> toward diastereoselective syntheses of **2** or **3** led by Bringmann,<sup>4a–c</sup> Lipshutz,<sup>4d,e</sup> Uemura,<sup>4f,g</sup> and Hoye.<sup>4h–k</sup> Notable strategies in constructing the axial chirality include kinetic asymmetric transformation of “lactone”,<sup>5</sup> atropselective intermolecular biaryl coupling,<sup>4e</sup> and chiral chromium complexes.<sup>6</sup> Despite these methods, an efficient route for catalytic asymmetric

syntheses of both **2** and **3** is still lacking. Consequently, a concise and stereoselective synthesis of **1** is yet to be reported.

While various methods for constructing chiral biaryls are available including chiral auxiliaries,<sup>7</sup> asymmetric transformations of “pro-chiral” biaryls,<sup>5,8</sup> asymmetric aryl formations,<sup>9</sup> asymmetric oxidative homocouplings,<sup>10</sup> the catalytic asymmetric cross-couplings,<sup>11</sup> particularly Suzuki-Miyaura couplings,<sup>12</sup> remains one of most attractive and versatile methods due to its mild reaction conditions, the ability of tolerating various functionalities, and the availability and nontoxic nature of starting materials. Despite the recent advances in this field, the current substrate scope of asymmetric Suzuki-Miyaura couplings remains very narrow. Limited results were achieved on substrates with synthetically interesting functionalities. To our knowledge, efficient asymmetric Suzuki-Miyaura coupling is yet to be applied in natural product syntheses. In addition, most current methods still suffer from high catalyst loadings and prolonged reaction time. We herein report an asymmetric Suzuki-Miyaura coupling methodology for chiral *ortho*-biaryl-ol derivatives with chiral monophosphorus ligands (Figure 2) and applications in total synthesis of **1–3**.

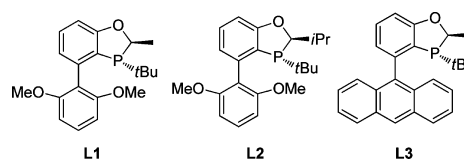


Figure 2. Chiral monophosphorus ligands.

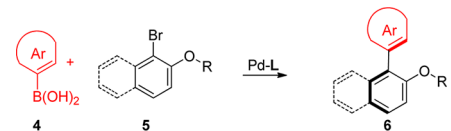
Although enantioselective Suzuki-Miyaura couplings are achieved on a few functionalized substrates with alkoxy,<sup>12e,f</sup> phosphonate,<sup>12a,h,i</sup> bulky amides,<sup>12b</sup> and carbonylbenzoxazolidinone<sup>13</sup> at *ortho* positions, it is particularly important to develop a highly efficient method for synthesizing chiral *ortho*-hydroxyl biaryls because such structures exist ubiquitously in numerous natural products. We thus focused on developing efficient asymmetric Suzuki-Miyaura couplings of aryl halides with an *ortho*-OPG group. Our previous work<sup>13</sup> on asymmetric Suzuki-Miyaura coupling with a Pd-L1 catalyst showed the importance of a second interaction between two coupling partners for high enantioselectivity. We thus exploited this strategy to explore the enantioselective Suzuki-Miyaura couplings between O-protected




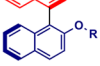


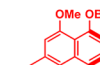
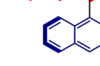
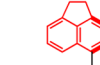
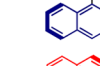
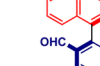
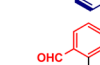
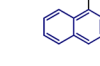
Received: September 18, 2013

Published: October 22, 2013

1-bromonaphthalen-2-ol and naphthalen-1-ylboronic acid. The reactions were carried out under nitrogen in toluene/water (5/1 v/v) with  $K_3PO_4$  as the base at 30–35 °C for 12 h (Table 1). It was found that the PG group exerts a significant impact on the enantioselectivity. While excellent yields (90–97% yield) were all obtained with various PG groups, only medium ee's (65–74%) were achieved with -CO $t$ Bu, -COPh, -CO(1-Np), and -P(O)Ph $_2$  groups (entries 1–3, 5). The carbonyl-oxazolidinone

**Table 1. Asymmetric Suzuki-Miyaura Coupling**

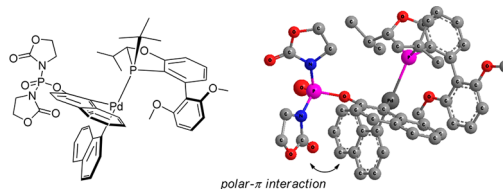


entry <sup>a</sup>	product <sup>b</sup>	L*	yield (%)	ee (%)
1	 (6a)	L1	97	72
2	 (6b)	L1	95	76
3	 (6c)	L1	96	76
4	 (6d)	L1	90	65
5	 (6e)	L1	97	72
6	 (6f)	L1	95	94
7	 (6g)	L2	96	99
8	 (6h)	L2	95	93
9	 (6i)	L2	94	92
10	 (6j)	L2	98	91
11	 (6k)	L1	95	90 <sup>c</sup>
12	 (6l)	L1	91	90
13	 (6m)	L3	95	95 <sup>d</sup>

<sup>a</sup>Conditions unless otherwise specified: 1.0 equiv of aryl bromide, 1.5 equiv of boronic acid, 1 mol % Pd(OAc) $_2$ , 1.2 mol % of chiral ligand, 3 equiv of  $K_3PO_4$ , toluene/H $_2$ O (5/1, v/v), rt, 4–12 h, isolated yield, enantiomeric excesses were determined by chiral HPLC on a chiralcel OD-H or chiralPak AD-H column. <sup>b</sup>The absolute configurations of 6a–f were determined by conversion to 1,1'-binaphthalen-2-ol and by comparison of their optical rotations to reported data. The absolute configurations of 6g–i were assigned by analogy. The absolute configurations of 6j–l were not determined. <sup>c</sup>BuOH/H $_2$ O (4/1, v/v) was employed the solvent and  $K_2CO_3$  was used as the base. <sup>d</sup>BuOH/H $_2$ O (4/1, v/v) was employed the solvent.

protection also led to only 65% ee (entry 4). Interestingly, an excellent enantioselectivity (94% ee) was achieved when a bis(2-oxo-3-oxazolidinyl)phosphinyl (BOP) protection was applied (entry 6). Further employment of the new ligand L2, with an isopropyl group on the top aryl ring, led to the coupling product 6f in an almost perfect ee (99%). Exploration of the substrate scope found that this protocol is tolerable with various substituents and functionalities on both coupling partners (entries 8–13). A variety of chiral functionalized biaryls could be efficiently synthesized in excellent ee's and yields under mild conditions by using these chiral monophosphorus ligands.

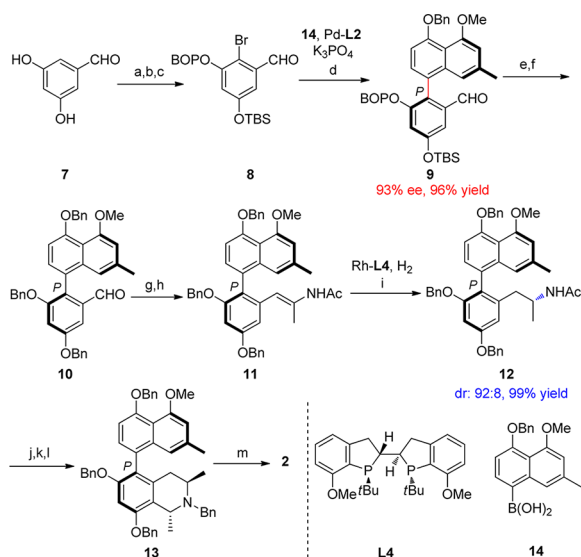
We attributed to a possible noncovalent interaction between two coupling partners for the high stereoselectivity. The sharp increase of enantioselectivity from OP(O)Ph $_2$  (entry 5) to OBOP (entry 6) functionality indicated the important effect posed by the oxazolidinone moieties in BOP. Observation from its substrate scope showed the necessity of an extended  $\pi$  system of arylboronic acid for the high enantioselectivity. For example, excellent enantioselectivities were achieved with 1-naphthylboronic acid, 2-formylphenylboronic acid, and *ortho*-phenylphenylboronic acid as shown in Table 1. In contrast, a poor selectivity (20% ee) was observed with 2-methylphenylboronic acid. It was thus proposed the presence of a significant polar- $\pi$  interaction<sup>14,15</sup> between the highly polarized BOP group and the extended  $\pi$  system of the boronic acid partner during reductive elimination step (Figure 3).



**Figure 3.** A proposed model of Pd(II)-L2 catalyst for the synthesis of 6f during the reductive elimination step.

The facile construction of a series of functionalized chiral biaryls under mild conditions by using this methodology prompted us to pursue the syntheses of both 2 and 3 through catalytic asymmetric Suzuki-Miyaura couplings. Because of the diastereomeric nature of 2 and 3 containing both axial chirality and central carbon chirality, a key challenge is to install the chiral elements in highly selective fashion and without interfering with each other. Our strategy is to construct the axial chirality using the developed asymmetric Suzuki-Miyaura coupling methodology followed by asymmetric hydrogenation to furnish the chiral tetrahydroisoquinoline moiety (Scheme 1).<sup>16</sup> Thus, diol 7 was sequentially protected with TBS and BOP followed by bromination to provide 8. Asymmetric Suzuki-Miyaura coupling of 8 with arylboronic acid 14<sup>17</sup> in the presence of 1 mol % Pd(OAc) $_2$  and 1.2 mol % L2 in toluene/water at 35 °C for 8 h provided the coupling product 9 in 96% yield and 93% ee at a gram scale. Subsequent basic removal of both BOP and TBS groups followed by benzyl protection yielded 10, which was further transformed to enamide 11 through nitroaldol and reductive acetylation.<sup>18</sup> Asymmetric hydrogenation of 11 with a Rh-WingPhos catalyst developed in our group<sup>19</sup> unfortunately led to the formation of 12 at a low d.r. ratio (2:1). A Rh-MeO-BIBOP catalyst<sup>20</sup> also provided a mediocre ratio (3:1). Further engineering of the hydrogenation catalyst led to the development of L4<sup>21</sup> with a chiral bisphosphoindole structure, providing 12 at a good d.r. ratio of 92:8 in 99% overall yield. Subsequent

## Scheme 1. Syntheses of 2



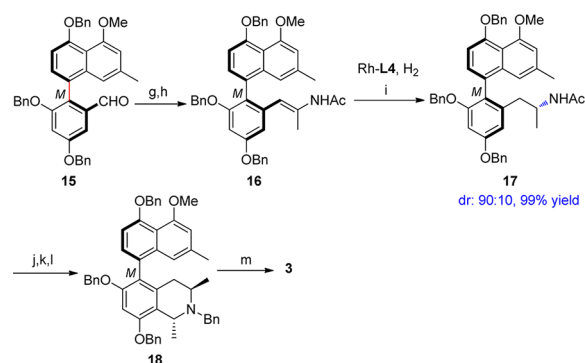
Conditions: (a) TBSCl, TEA, DCM,  $-78$  °C to rt, 59%; (b) NBS, DCM,  $-15$  °C, 86%; (c) BOPCl, TEA, DCM, 98%; (d) **14**, Pd(OAc)<sub>2</sub>/L2, K<sub>3</sub>PO<sub>4</sub>, toluene/H<sub>2</sub>O = 5/1, 96%, 93% ee; (e) NaOH, MeOH, 97%; (f) BnBr, K<sub>2</sub>CO<sub>3</sub>, DMF, 95%; (g) EtNO<sub>2</sub>, HO(CH<sub>2</sub>)<sub>2</sub>NH<sub>3</sub><sup>+</sup>HCO<sub>2</sub><sup>-</sup>, 60 °C, 92%; (h) Fe, AcOH, Ac<sub>2</sub>O, DMF, 70 °C, 60%; (i) 1 mol % Rh(nbd)<sub>2</sub>BF<sub>4</sub>/L4, H<sub>2</sub> (20 atm), DCM, 0 °C, 12 h, 99%, d.r. 92/8; (j) BnBr, NaH, DMF, 96%; (k) POCl<sub>3</sub>, 2,4,6-collidine, toluene, 100 °C, 4 h; (l) NaBH<sub>4</sub>, EtOH,  $-78$  °C to rt, 85% (over 2 steps, *trans*:*cis* = 6/1); (m) Pd/C (5%), H<sub>2</sub> (1 atm), MeOH/DCM = 2/1, 3 h, 95%.

installation of the tetrahydroquinoline moiety was accomplished through the following transformations: (i) benzyl protection at the amide bond under conditions of NaH/BnBr; (ii) Bischler-Napieralski reaction with POCl<sub>3</sub>; and (iii) diastereoselective reduction of the resulting imine with NaBH<sub>4</sub> (*trans*:*cis* = 6:1). A final global debenzoylation by hydrogenolysis provided **2**, whose characterization was identical at every aspects with reported data.<sup>2,4</sup> Thus, **2** was synthesized through an efficient asymmetric Suzuki-Miyaura coupling and an effective diastereoselective hydrogenation as key steps from compound **7** in 24% overall yield and in 13 linear steps

By using a similar route, the synthesis of **3** was also accomplished through asymmetric catalytic technology (Scheme 2). Thus, chiral aldehyde **15** with *M* configuration was synthesized from **8** and **14** with a Pd-*ent*-L2 catalyst in 96% yield and in 93% ee. After two steps to install the enamide moiety, asymmetric hydrogenation of **16** in the presence of 1 mol % Rh-L4 catalyst provided **17** with a d.r. ratio of 90:10. The fact that similar diastereoselectivities were observed in preparation of both **17** and **13** strongly demonstrated the independence of asymmetric hydrogenation with the Rh-L4 catalyst from the axial chirality of the substrate. **17** underwent installation of the tetrahydroquinoline moiety and global deprotection of the benzyl groups, yielding **3** in equal synthetic efficiency.

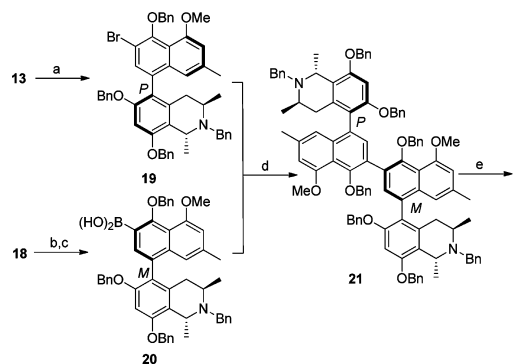
With the syntheses of both korupensamine A and B accomplished, we set out to synthesize michellamine B<sup>22</sup> (**1**) from their advanced intermediates **13** and **18** by a Suzuki-Miyaura coupling (Scheme 3).<sup>22b</sup> Thus, bromide **19** was prepared from **13** under action of C<sub>5</sub>H<sub>5</sub>NHBr<sub>3</sub> as the bromination reagent. Boronic acid **20** was obtained from **18** by bromination followed by metal-halogen exchange to react with B(OiPr)<sub>3</sub>. Suzuki-Miyaura coupling of **19** and **20** with Pd(PPh<sub>3</sub>)<sub>4</sub>

## Scheme 2. Syntheses of 3



Conditions: (g) EtNO<sub>2</sub>, HO(CH<sub>2</sub>)<sub>2</sub>NH<sub>3</sub><sup>+</sup>HCO<sub>2</sub><sup>-</sup>, 60 °C, 93%; (h) Fe, AcOH, Ac<sub>2</sub>O, DMF, 70 °C, 61%; (i) 1 mol % [Rh(NBD)<sub>2</sub>]BF<sub>4</sub>/L4, H<sub>2</sub> (20 atm), 0 °C, 12 h, 99%, d.r. = 90/10; (j) BnBr, NaH, DMF, 96%; (k) POCl<sub>3</sub>, 2,4,6-Collidine, toluene, 100 °C, 4 h; (l) NaBH<sub>4</sub>, EtOH,  $-78$  °C to rt, 78% (over 2 steps, *trans*:*cis* = 4/1); (m) Pd/C (5%), H<sub>2</sub> (1 atm), 3 h, MeOH/DCM = 2/1, 3 h, 86%.

## Scheme 3. Syntheses of 1



Conditions: (a) C<sub>5</sub>H<sub>5</sub>NHBr<sub>3</sub>, CHCl<sub>3</sub>, AcOH, 59%; (b) C<sub>5</sub>H<sub>5</sub>NHBr<sub>3</sub>, CHCl<sub>3</sub>, AcOH, 62%; (c) B(OiPr)<sub>3</sub>, toluene/THF = 2/1,  $-95$  °C, *n*BuLi,  $-95$  °C to  $-15$  °C, aq NH<sub>4</sub>Cl (87%); (d) Pd(PPh<sub>3</sub>)<sub>4</sub>, K<sub>3</sub>PO<sub>4</sub>, DMF, 90 °C, 84%; (e) Pd/C (5%), H<sub>2</sub> (1 atm), MeOH/DCM = 2/1, 7 h, 93%.

as the catalyst led to the formation of the coupling product **21** in 84% yield. Hydrogenolysis of **21** removed all benzyl groups and completed the synthesis of **1** in a concise and highly stereoselective fashion.

In summary, we have developed an efficient and practical asymmetric Suzuki-Miyaura coupling methodology for a series of synthetically useful and functionalized chiral biaryls under mild reaction conditions and at low catalyst loadings. The presence of a polar- $\pi$  interaction between the highly polarized BOP group and the extended  $\pi$  system of arylboronic acid coupling partner is believed to be essential for the high enantioselectivity. The methodology has been successfully applied in natural product syntheses and allowed for the first time the syntheses of both korupensamine A and B through catalytic asymmetric Suzuki-Miyaura coupling along with an effective diastereoselective hydrogenation. Additionally, the synthesis of their heterodimer **1** is accomplished in a concise and highly stereoselective fashion. Further applications of this methodology in constructing chiral biaryl natural products and analogs are currently underway and progress will be reported in due course.

## ■ ASSOCIATED CONTENT

## ■ Supporting Information

Full experimental details and characterization data. This information is available free of charge via the Internet at <http://pubs.acs.org/>.

## ■ AUTHOR INFORMATION

## Corresponding Author

tangwenjun@sioc.ac.cn

## Notes

The authors declare no competing financial interest.

## ■ ACKNOWLEDGMENTS

This work is supported by NSFC-21272254, STCSM-13PJ1410900, the "Thousand Plan" Youth program, the Innovation Fund of State Key Laboratory of Bio-organic and Natural Products Chemistry, and Boehringer Ingelheim. We are grateful to Dr. Lifeng Pan for NMR assistance.

## ■ REFERENCES

- (1) (a) Manfredi, K. P.; Blunt, J. W.; Cardellina, J. H., II; McMahon, J. B.; Pannell, L. L.; Cragg, G. M. *J. Med. Chem.* **1991**, *34*, 3402. (b) Boyd, M. R.; Hallock, Y. F.; Cardellina, J. H., II; Manfredi, K. P.; Blunt, J. W.; McMahon, J. B.; Buckheit, R. W.; Bringmann, G.; Schaffer, M.; Cragg, G. M.; Thomas, D. W.; Jato, J. G. *J. Med. Chem.* **1994**, *37*, 1740. (c) Hallock, Y. F.; Manfredi, K. P.; Dai, J. R.; Cardellina, J. H., II; Gulakowski, R. J.; McMahon, J. B.; Schaffer, M.; Stahl, M.; Gulden, K. P.; Bringmann, G.; Francois, G.; Boyd, M. R. *J. Nat. Prod.* **1997**, *60*, 677. (d) Bringmann, G.; Pokorny, F. In *The Alkaloids*; Cordell, G. A., Ed.; Academic Press: New York, 1995; Vol. 46, pp 127.
- (2) (a) Hallock, Y. F.; Manfredi, K. P.; Blunt, J. W.; Cardellina, J. H., II; Schäffer, M.; Gulden, K.-P.; Bringmann, G.; Lee, A. Y.; Clardy, J.; François, G.; Boyd, M. R. *J. Org. Chem.* **1994**, *59*, 6349. (b) Bringmann, G.; Gulden, K.-P.; Hallock, Y. F.; Manfredi, K. P.; Cardellina, J. H., II; Boyd, M. R.; Kramer, B.; Fleischhauer, J. *Tetrahedron* **1994**, *50*, 7807.
- (3) Reviews on syntheses of chiral biaryls: (a) Bringmann, G.; Gulder, T.; Gulder, T. A. M.; Breuning, M. *Chem. Rev.* **2011**, *111*, 563. (b) Chang, J.; Reiner, J.; Xie, J. *Chem. Rev.* **2005**, *105*, 4581. (c) Bringmann, G.; Mortimer, A. J. P.; Keller, P. A.; Gresser, M. J.; Garner, J.; Breuning, M. *Angew. Chem., Int. Ed.* **2005**, *44*, 5384.
- (4) (a) Bringmann, G.; Götz, R.; Keller, P. A.; Walter, R.; Henschel, P.; Schaffer, M.; Stablein, M.; Kelly, T. R.; Boyd, M. R. *Heterocycles* **1994**, *39*, 503. (b) Bringmann, G.; Ochse, M. *Synlett* **1998**, 1294. (c) Bringmann, G.; Ochse, M.; Götz, R. *J. Org. Chem.* **2000**, *65*, 2069. (d) Lipshutz, B. H.; Keith, J. M. *Angew. Chem., Int. Ed.* **1999**, *38*, 3530. (e) Huang, S.; Petersen, T. B.; Lipshutz, B. H. *J. Am. Chem. Soc.* **2010**, *132*, 14021. (f) Watanabe, T.; Shakadou, M.; Uemura, M. *Synlett* **2000**, 1141. (g) Watanabe, T.; Tanaka, Y.; Shoda, R.; Sakamoto, R.; Kamikawa, K.; Uemura, M. *J. Org. Chem.* **2004**, *69*, 4152. (h) Hoye, T. R.; Chen, M. *J. Org. Chem.* **1996**, *61*, 7940. (i) Hoye, T. R.; Chen, M. *Tetrahedron Lett.* **1996**, *37*, 3099. (j) Hoye, T. R.; Chen, M. *Tetrahedron Lett.* **1996**, *37*, 3097. (k) Hoye, T. R.; Chen, M.; Hoang, B.; Mi, L.; Priest, O. P. *J. Org. Chem.* **1999**, *64*, 7184.
- (5) (a) Bringmann, G.; Menche, D. *Acc. Chem. Res.* **2001**, *34*, 615. (b) Bringmann, G.; Breuning, M.; Tasler, S. *Synthesis* **1999**, 4, 525.
- (6) Kamikawa, K.; Uemura, M. *Synlett* **2000**, 938.
- (7) (a) Lipshutz, B. H.; Kayser, F.; Liu, Z.-P. *Angew. Chem., Int. Ed. Engl.* **1994**, *33*, 1842. (b) Chau, P.; Czuba, I. R.; Rizzacasa, M. A.; Bringmann, G.; Gulden, K.-P.; Schäffer, M. *J. Org. Chem.* **1996**, *61*, 7101. (c) Meyers, A. I.; Willemsen, J. J. *Tetrahedron* **1998**, *54*, 10493. (d) Meyers, A. I. *J. Heterocycl. Chem.* **1998**, *35*, 991.
- (8) (a) Hayashi, T.; Niizuma, S.; Kamikawa, T.; Suzuki, N.; Uozumi, Y. *J. Am. Chem. Soc.* **1995**, *117*, 9101. (b) Gustafson, J. L.; Lim, D.; Miller, S. *J. Science* **2010**, *328*, 1251. (c) Mori, K.; Ichikawa, Y.; Kobayashi, M.; Shibata, Y.; Yamanaka, M.; Akiyama, T. *J. Am. Chem. Soc.* **2013**, *135*, 3964.
- (9) (a) Gutnov, A.; Heller, B.; Fischer, C.; Drexler, H.-J.; Spannenberg, A.; Sundermann, B.; Sundermann, C. *Angew. Chem., Int. Ed.* **2004**, *43*, 3795. (b) Nishida, G.; Noguchi, K.; Hirano, M.; Tanaka, K. *Angew. Chem., Int. Ed.* **2007**, *46*, 3951. (c) Shibata, T.; Fujimoto, T.; Yokota, K.; Takagi, K. *J. Am. Chem. Soc.* **2004**, *126*, 8382.
- (10) (a) Mulrooney, C. A.; DiVirgilio, E. S.; Kozłowski, M. C. *J. Am. Chem. Soc.* **2003**, *125*, 6856. (b) Guo, Q.-X.; Wu, Z.-J.; Luo, Z.-B.; Liu, Q.-Z.; Ye, J.-L.; Luo, S.-W.; Cun, L.-F.; Gong, L.-Z. *J. Am. Chem. Soc.* **2007**, *129*, 13927. (c) Egami, H.; Katsuki, T. *J. Am. Chem. Soc.* **2009**, *131*, 6082.
- (11) Asymmetric Kumada couplings: (a) Hayashi, T.; Hayashizaki, K.; Kiyoi, T.; Ito, Y. *J. Am. Chem. Soc.* **1988**, *110*, 8153. (b) Hayashi, T.; Niizuma, S.; Kamikawa, T.; Suzuki, N.; Uozumi, Y. *J. Am. Chem. Soc.* **1995**, *117*, 9101. Asymmetric Nigishi couplings: (c) Genov, M.; Fuentes, B.; Espinet, P.; Pelaz, B. *Tetrahedron: Asymmetry* **2006**, *17*, 2593. (d) Genov, M.; Almorin, A.; Espinet, P. *Tetrahedron: Asymmetry* **2007**, *18*, 625.
- (12) (a) Yin, J. J.; Buchwald, S. L. *J. Am. Chem. Soc.* **2000**, *122*, 12051. (b) Shen, X.; Jones, G. O.; Watson, D. A.; Bhayana, B.; Buchwald, S. L. *J. Am. Chem. Soc.* **2010**, *132*, 11278. (c) Cammidge, A. N.; Crépy, K. V. L. *Chem. Commun.* **2000**, 1723. (d) Cammidge, A. N.; Crépy, K. V. L. *Tetrahedron* **2004**, *60*, 4377. (e) Bermejo, A.; Ros, A.; Fernández, R.; Lassaletta, J. M. *J. Am. Chem. Soc.* **2008**, *130*, 15798. (f) Uozumi, Y.; Matsuura, Y.; Arakawa, T.; Yamada, Y. M. A. *Angew. Chem., Int. Ed.* **2009**, *48*, 2708. (g) Sawai, K.; Tatum, R.; Nakhodo, T.; Fujihara, H. *Angew. Chem., Int. Ed.* **2008**, *47*, 6917. (h) Zhang, S.-S.; Wang, Z. Q.; Xu, M.-H.; Lin, G.-Q. *Org. Lett.* **2010**, *12*, 5546. (i) Genov, M.; Almorin, A.; Espinet, P. *Chem.—Eur. J.* **2006**, *12*, 9346. (j) Yamamoto, T.; Akai, Y.; Nagata, Y.; Sugino, M. *Angew. Chem., Int. Ed.* **2011**, *50*, 8844. (k) Wang, S.; Li, J.; Miao, T.; Wu, W.; Li, Q.; Zhuang, Y.; Zhou, Z.; Qiu, L. *Org. Lett.* **2012**, *14*, 1966.
- (13) Tang, W.; Patel, N. D.; Xu, G.; Xu, X.; Savoie, J.; Ma, S.; Hao, M.-H.; Keshipeddy, S.; Capacci, A. G.; Wei, X.; Zhang, Y.; Gao, J. J.; Li, W.; Rodriguez, S.; Lu, B. Z.; Yee, N. K.; Senanayake, C. H. *Org. Lett.* **2012**, *14*, 2258.
- (14) (a) Cozzi, F.; Cinquini, M.; Annuziata, R.; Siegel, J. S. *J. Am. Chem. Soc.* **1993**, *115*, 5330. (b) Chen, C.-T.; Siegel, J. S. *J. Am. Chem. Soc.* **1994**, *116*, 5959. (c) Pace, C. J.; Gao, J. *Acc. Chem. Res.* **2013**, *46*, 907.
- (15) Recent review on various aromatic interactions: Salonen, L. M.; Ellermann, M.; Diederich, F. *Angew. Chem., Int. Ed.* **2011**, *50*, 4808.
- (16) A strategy by first installation of the chiral tetrahydroisoquinoline moiety followed by atropselective biaryl construction was also executed. High atropselectivity (d.r. 97:3) was achieved for the synthesis of korupesamine B, while poor selectivity (d.r. 65:35) was observed for the synthesis of korupesamine A. See SI for details.
- (17) See SI for a concise synthesis of **14**.
- (18) Laso, N. M.; Quiclet-Sire, B.; Zard, S. Z. *Tetrahedron Lett.* **1996**, *37*, 1605.
- (19) Liu, G.; Liu, X.; Cai, Z.; Jiao, G.; Xu, G.; Tang, W. *Angew. Chem., Int. Ed.* **2013**, *52*, 4235.
- (20) (a) Tang, W.; Qu, B.; Capacci, A. G.; Rodriguez, S.; Wei, X.; Haddad, N.; Narayanan, B.; Ma, S.; Grinberg, N.; Yee, N. K.; Krishnamurthy, D.; Senanayake, C. H. *Org. Lett.* **2010**, *12*, 176. (b) Li, W.; Rodriguez, S.; Duran, A.; Sun, X.; Tang, W.; Premasiri, A.; Wang, J.; Sidhu, K.; Patel, N. D.; Savoie, J.; Qu, B.; Lee, H.; Haddad, N.; Lorenz, J. C.; Nummy, L.; Hossain, A.; Yee, N.; Lu, B.; Senanayake, C. H. *Org. Process. Res. Dev.* **2013**, *12*, 1061–1065.
- (21) See SI for a detailed synthesis.
- (22) (a) Bringmann, G.; Götz, R.; Keller, P. A.; Walter, R.; Henschel, P.; Schaffer, M.; Stablein, M.; Kelly, T. R.; Boyd, M. R. *Heterocycles* **1994**, *39*, 503. (b) Hobbs, P. D.; Upender, V.; Liu, J.; Pollart, D. J.; Thomas, D. W.; Dawson, M. I. *Chem. Commun.* **1996**, 923. (c) Bringmann, G.; Götz, R.; Keller, P. A.; Walter, R.; Boyd, M. R.; Lang, F.; Garcia, A.; Walsh, J. J.; Tellitu, I.; Bhaskar, K. V.; Kelly, T. R. *J. Org. Chem.* **1998**, *63*, 1090.